

## Equations of State

For non ideal gases we will need a relationship for work and this requires a relationship for PVT (an equation of state):

What are the assumptions for an Ideal Gas and their interactions ?

How will real gases differ ?

Q and T (above)

V for T and P

Many first approximations start with a compressibility factor, Z:

$PV = nZRT$  where for one mole:  $Z \frac{PV}{RT} =$  function of ‘

$$Z = 1 + B'P + C'P^2 + D'T + \dots \text{ or } = \frac{B}{V} + \frac{C}{V^2} + \dots$$

or  $Z = F(T_r, P_r)$ ; one example  $Z^0 + Z^1$ ; for gas (tables)

or [Virial]  $PV = a + bP + cP^2 + \dots$

or van der Waals

$$P = \frac{RT}{V - b} - \frac{a}{V^2}; \text{ a and b for gas (tables)}$$

*Van der Waals equation*

$$\left(P + \frac{a}{\tilde{V}^2}\right)(\tilde{V} - b) = RT$$

$$P = \frac{RT}{(\tilde{V} - b)} - \frac{a}{\tilde{V}^2}$$

Parameters for the Van der Waals Equation

Gas	$a, Pa \cdot m^6/mol^2$	$b, m^3/mol \times 10^3$
O <sub>2</sub>	0.1381	3.184
N <sub>2</sub>	0.1368	3.864
H <sub>2</sub> O	0.5542	3.051
CH <sub>4</sub>	0.2303	4.306
CO	0.1473	3.951
CO <sub>2</sub>	0.3658	4.286
NH <sub>3</sub>	0.4253	3.737
H <sub>2</sub>	0.0248	2.660
He	0.00346	2.376

*Redlich-Kwong Equation*

$$P = \frac{RT}{(\tilde{V} - b)} - \frac{a}{T^{1/2} \tilde{V}(\tilde{V} + b)}$$

*Peng-Robinson Equation*

$$P = \frac{RT}{(\tilde{V} - b)} - \frac{a(T)}{\tilde{V}(\tilde{V} + b) + b(\tilde{V} - b)}$$

*Generalized Cubic Equation*

$$P = \frac{RT}{(\tilde{V} - b)} - \frac{(\tilde{V} - c)}{(\tilde{V} + d)(\tilde{V}^2 - \tilde{V} + e)}$$

Note that if  $Z = PV/NRT$ , the equation can be written as a cubic equation in  $Z$

$$Z^3 + Z^2 + Z + \dots = 0$$

### Solution for Selected Equations of State

The usual cubic equations of state can all be expressed in the same form:

$$Z^3 + Z^2 + Z + \dots = 0$$

For the van der Waals, Redlich-Kwong, and Peng-Robinson equations, the Table below gives the relationship between the parameters  $a$ ,  $b$ ,  $c$ ,  $d$ ,  $e$  and the parameters for the respective equations of state.

	<i>van der Waals</i>	<i>Redlich-Kwong</i>	<i>Peng-Robinson</i>
	$-1 - B$	$-1$	$-1 + B$
	$A$	$A - B - B^2$	$A - 3B^2 - 2B$
	$-AB$	$-AB$	$-AB + B^2 + B^3$

where

$$Z = \frac{P \tilde{V}}{RT} \quad \text{and} \quad B = \frac{P b}{RT}$$

$$A = \frac{aP}{(RT)^2} \quad \text{for van der Waals and Peng-Robinson}$$

$$A = \frac{aP}{(RT)^2 \sqrt{T}} \quad \text{for Redlich-Kwong}$$

# Peng-Robinson Equation of State

The cubic form of the Peng-Robinson equation is the following

$$Z^3 + Z^2 + Z + = 0$$

noting that the definitions of  $Z$ ,  $A$ , and  $B$  are

$$Z = \frac{P \tilde{V}}{R T}$$

Using the generalized form for the parameters in the Peng-Robinson Equation, we obtain a simplified set of parameters to use in the equation

$$A = \frac{0.45724P_r}{T_r^2} (T_r) \quad B = 0.07780 \frac{P_r}{T_r}$$

with other parameters appropriately defined

$$\begin{aligned} (T_r) &= 1 + (1 - \sqrt{T_r}) \\ &= 0.37464 + 1.5422 - 0.26992 \end{aligned}$$

So that the parameters in the cubic form are given by

$$\begin{aligned} &= -1 + 0.07780 \frac{P_r}{T_r} \\ &= \frac{0.45724P_r}{T_r^2} (T_r) - 2 \left( 0.07780 \frac{P_r}{T_r} \right) - 3 \left( 0.07780 \frac{P_r}{T_r} \right)^2 \\ &= \left( 0.07780 \frac{P_r}{T_r} \right) \frac{0.45724P_r}{T_r^2} (T_r) + \left( 0.07780 \frac{P_r}{T_r} \right)^3 + \left( 0.07780 \frac{P_r}{T_r} \right)^2 \end{aligned}$$

To relate the two types of Representation:

$$P = \frac{RT}{v - b} - \frac{a}{v(v+b) + b(v-b)} \text{ where } b = \frac{BRT}{P} \text{ \& } a = \frac{A(RT)^2}{P}$$

$$P = \frac{RT}{v - \frac{BRT}{P}} - \frac{\frac{A(RT)^2}{P}}{v\left(v + \frac{BRT}{P}\right) + \frac{BRT}{P}\left(v - \frac{BRT}{P}\right)}$$

gathering P in denominator

$$P = \frac{RTP}{Pv - BRT} - \frac{\frac{A(RT)^2}{P}}{\frac{v}{P}(Pv + BRT) + \frac{BRT}{P^2}(Pv - BRT)}$$

Multiply by V/RT and elim. /P in second term:

$$\frac{Pv}{RT} = \frac{Pv}{Pv - BRT} - \frac{A(RT)*Pv}{Pv(Pv + BRT) + BRT(Pv - BRT)}$$

Sub. for Z; top & bottom by 1/RT; top & bottom by 1/RTPV

$$Z = \frac{Z}{Z - B} - \frac{\frac{A}{RT}}{\frac{(Pv + BRT)}{RT} + B\frac{(Pv - BRT)}{Pv}} = \frac{Z}{Z - B} - \frac{A}{(Z + B) + B - \frac{B^2}{Z}}$$

$$Z = \frac{Z}{Z - B} - \frac{AZ}{Z^2 + 2BZ - B^2} \quad 1 = \frac{1}{Z - B} - \frac{A}{Z^2 + 2BZ - B^2}$$

$$0 = (Z - B)(Z^2 + 2BZ - B^2) - Z^2 - 2BZ + B^2 + A(Z - B)$$

$$0 = Z^3 + Z^2(-B + 2B - 1) + Z(-B^2 - 2B^2 + A - 2B) + B^3 + B^2 - AB$$

$$0 = Z^3 + Z^2(B - 1) + Z(A - 3B^2 - 2B) + B^3 + B^2 - AB$$

$$0 = Z^3 + Z^2 + Z + \quad : \text{where } = (B - 1); = (A - 3B^2 - 2B); = (B^3 + B^2 - AB)$$

One approach to the use of the Peng-Robinson:

using  $\omega$ ,  $T_c$  and  $P_c$

define

$$Z = 0.37464 + 1.54226\omega - 0.26992\omega^2$$

$$\sqrt{Z} = 1 + (1 - T_r^{1/2})$$

$$a = (0.45724) \frac{R^2 T_c^2}{P_c}$$

$$b = 0.0778 \frac{RT_c}{P_c}$$

$$P = \frac{RT}{v - b} - \frac{a}{v(v+b) + b(v-b)}$$

to iterate multiply by  $(v-b)/P$  and rearrange:

$$v = b + \frac{RT}{P} - \frac{a/P}{v \left( \frac{v+b}{v-b} \right) + b}$$

calculate  $\left( \frac{v+b}{v-b} \right)$  then  $v \left( \frac{v+b}{v-b} \right) + b$  then  $\frac{a/P}{v \left( \frac{v+b}{v-b} \right) + b}$

This then gives  $v$ , the next guess

Proceed until  $v$  does not vary

## The solution of a Cubic Equation

If a cubic equation is stated as below :

$$Z^3 + Z^2 + Z + = 0$$

its roots can be obtained if we examine the following forms :

$$q = \frac{1}{3} - \frac{1}{9} \quad ; \quad r = \frac{1}{6} ( - 3 ) - \frac{1}{27} \quad 3$$

If  $q^3 + r^2 > 0$  , there will be one real root and a pair of complex conjugate roots.

If  $q^3 + r^2 = 0$  , all roots are real and at least two will be equal.

If  $q^3 + r^2 < 0$  , all roots are real ( irreducible case, i.e., no analytical solution)

The roots can be expressed using the following definitions :

$$s_1 = [r + 3\sqrt{(q^3 + r^2)}]^{1/3} \quad ; \quad s_2 = [r - 3\sqrt{(q^3 + r^2)}]^{1/3}$$

and the roots are :

$$Z_1 = (s_1 + s_2) - \frac{1}{3}$$

$$Z_2 = -\frac{(s_1 + s_2)}{2} - \frac{1}{3} + \frac{i\sqrt{3}}{2}(s_1 - s_2)$$

$$Z_3 = -\frac{(s_1 + s_2)}{2} - \frac{1}{3} - \frac{i\sqrt{3}}{2}(s_1 - s_2)$$

You should note some interesting and useful properties of the roots.

$$Z_1 + Z_2 + Z_3 = - \quad ; \quad Z_1Z_2 + Z_1Z_3 + Z_2Z_3 =$$

$$Z_1Z_2Z_3 = -$$

